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Concentration of Mg^{2+} / $\mu\text{g cm}^{-3}$ 2.00 4.00 6.00 8.00 10.00 Absorbance 0.085 0.169 0.242 0.316 0.393 Construct a calibration graph using the axes below. [2] Absorbance 0.40 0.35 0.30 0.25 0.20 0.15 0.10 0.05 0 0 2.0 4.0 6.0 8.0 10.0 Concentration of Mg^{2+} / $\mu\text{g cm}^{-3}$ (b) When a 0.200 g sample of cast iron was dissolved in acid to produce 100 ...

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Many recent developments in chemistry have involved making use of devices that operate on a nanoscale. (a) (i) State the scale at which nanotechnology takes place and outline the importance of

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3 2 PO B. NH 3 HCO 3 PO 4 3 C. NH 4 HCO 3 2 PO 4 2 D. NH 4 HCO 3 PO 4 3 12. What happens when magnesium metal reacts with chlorine gas? A. Each magnesium atom loses two electrons and each chlorine atom gains two electrons. B. Each magnesium atom gains one electron and each chlorine atom loses one electron. C.

Chemistry SL paper 1 TZ1 2009 - Exam

coefficient of 3 on stearic acid; [3] Accept displayed or condensed formulas for molecules (b) (i) both have first double bond on C9 with carbon / linoleic has an $\omega - 6$ C=C

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