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Guide

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Chapter 8: Solutions,
Acids, and Bases.

STUDY. PLAY. Solute.

Substance that is
dissolved in a solution

Ex. Oxygen in air.

Solvent. The substance

in which the solute

dissolves (substance

that "does" the

dissolving) Ex. Nitrogen

in air. Dissociation. The

process in which an

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ionic compound
separates into ions as
it dissolves.

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The Solutions, Acids,
and Bases chapter of
this Prentice Hall
Physical Science
Companion Course
helps students learn
the essential lessons
associated with acids,
bases and solutions.

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Each of these...

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Chapter 8--Solutions,
Acids, & Bases,

Chapter 8: Solutions,
Acids, and Bases.

Physical Science

vocabulary; Prentice

Hall; Chapter 8. STUDY.

PLAY. solute. The

substance that is

dissolved in a solvent

to make a solution.

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solvent. The substance in which a solute is dissolved to form a solution.

Chapter

8--Solutions, Acids,

& Bases, Chapter 8:

Solutions ...

Chapter 8 Solutions,
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Section 8.4 Strength of
Acids and Bases (pages

246–249) This section
explains how to

describe acids and

bases in terms of both

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Strategy (page 246)

Comparing and

Contrasting As you

read, complete the

diagram by comparing

and contrasting acids

and bases. For more

information on

Chapter 8 Solutions, Acids, and Bases Section 8.4 Strength

...

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Answers 16. For a solution to form, one substance must dissolve in another. For this to happen, the solute and solvent particles must ATTRACT ONE ANOTHER. 17. During the formation of a solution, energy is ABSORBED & RELEASED. 18. The solubility of solids increases as the temperature of the solvent INCREASES.

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**Ch 8 Solutions, Acids
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Guide Answers**

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16. For a solution to form, one substance must dissolve in another. For this to happen, the solute and solvent particles must _____. 17. During the formation of a solution, energy is _____. 18.

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Chapter 8 Solutions,
Acids, and Bases
Section 8.3 Properties
of Acids and Bases
(pages 240–245) This
section describes the
general properties of
acids and bases.

**Chapter 8 Solutions,
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Section 8.3 ...**

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12th grade. 14 times.

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average accuracy. 6

months ago. kallen. 0.

Save. Edit. Edit. ... If 20

cm³ samples of 0.1

mol dm⁻³ solutions of

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the acids below are taken, which acid would require a different volume of 0.1 mol dm⁻³ sodium hydroxide for complete neutralization?

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Chapter 8 Solutions, Acids, And Bases; matthew j. • 19 cards. solute. a substance whose particles are

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dissolved in a solution.
solvent. the substance
in which the solute
dissolves. dissociation.
the process in which an
ionic compound
separates into ions as
it dissolves. ...

chapter 8 solutions, acids, and bases - Physical Science ...

A chemical reaction
between an acid and
a base: Salt: An ionic
compound formed
when an acid reacts

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with a base: pH: A
measure of the
hydronium ion
concentration of a
solution: Buffer: A
solution that is
resistant to large
changes in pH:
Electrolyte: A
compound that
produces ions when it
dissolves

Quia - Chapter 8: Solutions, Acids, and Bases

Chapter 8, Solutions,
Page 17/25

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Solutions, Acids

Acids, and Bases.

Acid Bases Study

Particles that are dissolved in a solution.

The solution in which a solute will dissolve in.

The process in which an ionic compound separates into ions as it dissolves. Breaking down into small pieces that spread out through the water.

Quia - Chapter 8, Solutions, Acids, and Bases

Chapter 8 vocabulary

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(Solutions, Acids, and

Bases) 8.1- Formation

of Solutions Ionization

The process in which

neutral molecules gain

or lose electrons.

Solute A substance

whose particles are

dissolved in a solution.

Kool-Aid Dispersion The

breaking into the small

pieces that

Chapter 8

vocabulary

(Solutions, Acids,

and Bases) by ...

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solution. 8.3 Properties of Acids and Bases

Some general properties of acids include sour taste, reactivity with metals, and ability to produce color changes in indicators.

- An acid is a compound that produces hydronium ions (H_3O^+) when dissolved in water.

- An indicator is any substance that changes color in the presence of an acid or

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base.

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**Chapter 8 Solutions,
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Chapter 8 Solutions,
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Physical Science

Reading and Study

Workbook Chapter 8 89

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Section 8.1 Formation
of Solutions (pages
228–234) This section

explains the parts of a

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**Chapter 8:
Solutions, Acids, and
Bases**

MSBSHSE Class 8

Science Chapter 12:

Introduction to Acid

and Base Important

Textbook Questions

and Solutions You will

notice that some

substances have a

sweet taste, some are

bitter, some are sour or

astringent. Lemon,

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tamarind, vinegar or amla like substances acquire sour taste due to the presence of a typical compound in them.

MSBSHSE Class 8 Science Chapter 12: Introduction to Acid

...

a chemical reaction between an acid and a base: salt: an ionic compound formed when an acid reacts with a base: pH: a

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measure of the
hydronium ion
concentration of a
solution: buffer: a
solution that is
resistant to large
changes in pH:
electrolyte: a
compound that
produces ions when it
dissolves

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